



LOAD RELAXATION STUDIES OF A METALLIC GLASS

T.D. Hadnagy, D.J. Krenitsky, D.G. Ast, Che-Yu Li

Contract Number: NRO39-151100014-77-C-0546 N

Technical Report #1

September 1977



79 04 26 048

DISTRIBUTION STATEMENT

Approved for public release, Distribution Unlimited SECURITY CLASSIFICATION OF THIS PAGE (When Date Entered)

REPORT DOCUMENTATION PAGE	READ INSTRUCTIONS BEFORE COMPLETING FORM
REPORT NUMBER	3. RECIPIENT'S CATALOG NUMBER
Technical Report No. 1	
TITLE (and Substite)	5 TYPE OF REPORT & PERIOD COVERED
6 Load Relaxation Studies 9	Technical Progress Pepet .
of a Metallic Glass.	6. PERFORMING ORG. REPORT NUMBER
. AUTHOR(a)	B. CONTRACT OR GRANT NUMBER(+)
D.G. Ast, Che-Yu/Li	(14) TR-1
PERFORMING ORGANIZATION NAME AND ADDRESS	10. PROGRAM ELEMENT, PROJECT, TASK
Material Science & Eng.; Bard Hall Cornell University, Ithaca, NY 14853	NRO39-151
1. CONTROLLING OFFICE NAME AND ADDRESS	12. REPORT DATE
Metallurgy Branch	Sep 77
Office of Naval Research	13. NUMBER OF PAGES
Arlington, VA 22217 4. MONITORING AGENCY NAME & ADDRESS(II different from Controlling Office)	15. SECURITY CLASS. (of this report)
WONTONING ROCKET WATER TO THE STATE OF THE S	Unclassified
(12) 17p.	ISA. DECLASSIFICATION DOWNGRADING SCHEDULE
6. DISTRIBUTION STATEMENT (of this Report)	
Distribution of this document is unlimited	
	WEGEZZIAL ALL
	om Report)
	om Report) ### Walle Section #### Bed Section ###################################
	om Report) 2315 waite section pec surf mother
7. DISTRIBUTION STATEMENT (of the abstract entered in Block 20, if different fro	om Report) ### Walle Section #### Bed Section ###################################
7. DISTRIBUTION STATEMENT (of the abstract entered in Block 20, if different fro	om Report) PRE Name Section PRE SECTION UNITEDATION
7. DISTRIBUTION STATEMENT (of the abstract entered in Block 20, if different fro	om Report) ### Walle Section ### ################################
7. DISTRIBUTION STATEMENT (of the abstract entered in Block 20, if different fro	om Report) ### Walle Section ### ################################
7. DISTRIBUTION STATEMENT (of the abstract entered in Block 20, if different fro	om Report) ### Malle Section #### #### #### #### #### #### ####
7. DISTRIBUTION STATEMENT (of the abstract entered in Block 20, if different fro	om Report) ### Malle Section #### #### #### #### #### #### ####
7. DISTRIBUTION STATEMENT (of the abstract entered in Block 20, if different fro	om Report) ### Malle Section #### #### #### #### #### #### ####
7. DISTRIBUTION STATEMENT (of the abstract entered in Block 20, if different from the supplementary notes 9. KEY WORDS (Continue on reverse side if necessary and identify by block number	om Report) ###################################
7. DISTRIBUTION STATEMENT (of the abstract entered in Block 20, if different from the supplementary notes 9. KEY WORDS (Continue on reverse side if necessary and identify by block number	om Report) ###################################
DISTRIBUTION STATEMENT (of the abstract entered in Block 20, if different from the supplementary notes B. Supplementary notes D. KEY WORDS (Continue on reverse elde if necessary and identify by block number Metallic Glass, Stress relaxation	om Report) PRE PRE PRE PRE PRE PRE PRE PR
7. DISTRIBUTION STATEMENT (of the abstract entered in Block 20, if different from the supplementary notes 8. Supplementary notes 9. Key words (Continue on reverse elde if necessary and identify by block number, Metallic Glass, Stress relaxation	om Report) PRE PRE PRE PRE PRE PRE PRE PR
7. DISTRIBUTION STATEMENT (of the abstract entered in Block 20, if different from the supplementary notes 8. Supplementary notes 9. Key words (Continue on reverse side if necessary and identify by block number, Metallic Glass, Stress relaxation 10. ABSTRACT (Continue on reverse side if necessary and identify by block number) The stress relaxation of the Fe-Ni based metalli	ent section and se
7. DISTRIBUTION STATEMENT (of the abstract entered in Block 20, if different from the state of the abstract entered in Block 20, if different from the state of t	ic glass Fe ₄₀ Ni ₄₀ P ₁₄ B ₆ The relaxation consists
7. DISTRIBUTION STATEMENT (of the abstract entered in Block 20, if different from the supplementary notes 8. Supplementary notes 9. Key words (Continue on reverse side if necessary and identify by block number, Metallic Glass, Stress relaxation 10. ABSTRACT (Continue on reverse side if necessary and identify by block number) The stress relaxation of the Fe-Ni based metalli	ic glass Fe ₄₀ Ni ₄₀ P ₁₄ B ₆ The relaxation consists followed by a history

DD 1 JAN 73 1473

EDITION OF 1 NOV 65 IS OBSOLETE 5./N 0102-LF-014-6601

Unclassified.
SECURITY CLASSIFICATION OF THIS PAGE (When Dote Entered)

403 152

nt

LOAD RELAXATION STUDIES OF A METALLIC GLASS

T. D. Hadnagy, D. J. Krenisky, D. G. Ast and Che-Yu Li Department of Materials Science and Engineering Cornell University, Ithaca, New York 14353

This note reports experimental results of load relaxation studies of a commercial metallic glass (METGLAS 2826) as a function of temperature. The data suggests that metallic glasses exhibit deformation behavior with flow laws similar to those governing plastic deformation in crystalline solids. The lack of appreciable work hardening on annealed material and the identification of an anelastic component are also indicated by the experimental observation.

The deformation properties of metallic glasses were reviewed recently by Davis. He discussed the possibility that metallic glasses exhibit deformation properties governed by dislocation mechanisms and show little capacity for work hardening. Recently Murata and co-workers showed that non-elastic deformation in metallic glasses contains two separate components: a time dependent and recoverable anelastic component and a time dependent and non-recoverable plastic component. They argued, however, based on their observations that metallic glasses can be hardened by straining.

noad relaxation experiments have been extensively used in the investigation of non-elastic properties of crystalline solids in conjunction with the work on the development of a plastic equation of state. They have the advantage that they can be used to generate stress-strain rate data covering a wide range of strain rate while avoiding the occurrence of plastic instability in the

specimen. 5 This is particularly useful in the investigation of metallic glasses.

Load relaxation data of a large variety of crystalline solids have shown that the same flow laws for plastic deformation (stress-strain rate relations) apply. 5,6,7 The contribution of the anelastic deformation component can be identified by the failure of a portion of the stress-strain rate data to be described by the flow laws for plastic deformation. 5

Based on the work mentioned above a deformation model was developed for non-elastic deformation by using the state variable approach. In the present work the same model will be adopted to analyze the load relaxation data of metallic glasses.

EXPERIMENTS

The tensile specimen used was made from a commercial metallic glass ribbon [METGLAS TM 2826 (0.65x0.0023 in.)] by polishing with 220 grit emery-paper. It had a gage section of 2.2x0.006x0.15 cm. The load relaxation experiment was carried out by using a table model Instron testing system. After the temperature of the specimen and the testing system was stabilized, the specimen was loaded at an extension rate of .02 in/min (.056 cm/sec) to the desired stress level. The crosshead was then fixed and load relaxation resulted from the conversion of the combined elastic strain of the test system and the specimen to non-elastic strain of the specimen. The rate of load relaxation was related to the non-elastic strain rate of the specimen through the combined modulus of the test system and specimen. The experiment yielded data in the form of load vs. time which was analyzed to yield a log σ vs log ε (σ , stress and ¿, non-elastic strain rate) curve for a particular load relaxation run. Throughout this study the specimen deformed homogeneously without the formation of shear band.

RESULTS AND DISCUSSION

Typical load relaxation data obtained at 270°C in the form of log σ - log ε curves are shown in Figure 1. Each curve represents the data obtained from a load relaxation run. The specimen used was first annealed at 300° for 8 hours prior to the load relaxation experiment at 270°C. All the curves shown in Figure 1 were obtained on a single specimen, by re-loading the specimen after a load relaxation run to a higher stress level and continuing raising the stress of each subsequent run until failure. The length of each

load relaxation run ranged from a few hours to in excess of a day. The order of the runs are as given in Figure 1. The data of the first run and the third run are not shown due to alignment and temperature control problem respectively.

Each log o - log & curve suggests distinctly two types of behavior. The initial portion of the load relaxation data (high strain rate portion) shows an extremely high stress exponent (>100, the inverse of the slope of the log o - log curve). As the experiment proceeds (low strain rate portion), all the log o - log & data merge into a single curve and show a significantly lower stress exponent of about 4. These results are consistent with the deformation model based on the state variable approach and the experience gained on crystalline solids 3,4,5,6 such that the initial high strain rate portion of the relaxation data reflect the contribution of anelastic deformation and the low strain rate portion of the data show the stress-strain rate relation of plastic deforma-These results also suggest the lack of significant work hardening as well as the absence thermally induced structural change. Discussion will be made in the following to support these possibilities.

the non-elastic properties of the grain matrix of crystalline solids and has been tested extensively by using a variety of crystalline metals and alloys. 3,4,5,8 It consists of two branches in parallel. One of the branches includes an anelastic spring in series with a plastic element (a-element) which governs plastic deformation at high homologeous temperatues and/or low strain rates. The other branch contains a non-elastic friction element which

represents dislocation glide controlled processes and is important at low homologeous temperatures and/or high strain rates. The magnitude of the anelastic strain a is linearly related to the stress on the anelastic spring through its modulus. The non-elastic strain rate of the specimen $\dot{\epsilon}$ is represented by

$$\dot{\epsilon} = \dot{a} + \dot{a}$$
 (1)

where \dot{a} is the anelastic strain rate and \dot{a} is the plastic strain rate of the \dot{a} -element. The following relations are of present interest:

$$\ln(\sigma^*/\sigma_a) = (\dot{\epsilon}^*/\dot{\alpha})^{\lambda} \tag{2}$$

$$\dot{\bullet} = \dot{a} * (\sigma_f/M)^M \tag{3}$$

$$\sigma = \sigma_{\mathbf{a}} + \sigma_{\mathbf{f}} \tag{4}$$

$$\sigma_a = Ma$$
 (5)

where σ^* is the hardness parameter which measures the hardness state of the specimen, σ_a is the stress on the $\dot{\sigma}$ -element, σ_f is the stress on the element controlled by dislocation glide, $\dot{\epsilon}^*$ is the rate constant for the $\dot{\sigma}$ -element, $\dot{\sigma}^*$ is the rate constant for dislocation glide controlled element, σ is applied stress, M is the anelastic modulus and λ and M are two constants which determine the shape of the constant hardness (σ^*) log σ - log $\dot{\epsilon}$ curve.

The hardness parameter σ^* determines a unique $\log \sigma - \log \varepsilon$ curve and can be changed by work hardening or by thermally induced structural changes. Equations (2) and (3) above represents the $\dot{\sigma}$ -element and the dislocation glide controlled element respectively. Typical values of λ and M for crystalline metals and alloys are λ , near 0.15 and M, between 7.5-8. 3 ,4,8

According to the above model, during the initial loading of

the specimen, the non-elastic strain rate of the specimen reflects mostly the inelastic strain rate a. This is because the initially stored anelastic strain of a well annealed specimen is close to zero. As the crosshead is fixed, the initial portion of the load relaxation is still controlled by the anelastic strain rate a due to the fact that the build-up of stored anelastic strain is governed by Equation (3) and requires a finite time. Consequently the initial portion of the log o - log & data cannot be presented by either Equation (2) or (3). At longer relaxation times, the rate of load relaxation decreases and plastic deformation becomes important. For the interest of the present discussion, the shape of the low strain rate portion of the log o - log curve [Equation (2)] reflects the extremely high hardness (o*) value involved. During the later stages of the relaxation period the stored anelastic strain will decrease significantly with time as described by Equations (3) and (5). Thus during reloading and the portion of the load relaxation immediately following reloading the anelastic strain rate a will be controlling again.

If following reloading to a higher stress the specimen has not been hardened by deformation (σ^* unchanged), the $\log \sigma - \log \varepsilon$ data obtained after anelasticity has taken its toll, will coincide with those obtained in the previous load relaxation run within the error of the experiment. This is shown by the data in Figure 1. A more detailed analysis shows very small but systematic shifts of the order of the experimental error of the slope of these curves. It is noted that for crystalline solids, after reloading to a higher stress level, because of work hardening, invariably a substantially new $\log \sigma - \log \varepsilon$ curve will result. It is noted also the data shown in Figure 2 suggests the absence of a thermally in-

duced structural change.

The existence of both the anelastic component and the plastic component of deformation suggested above and the very low stress dependence for plastic flow reflected by the overlapping portion of the log σ - log $\dot{\epsilon}$ curves in Figure 1 are consistent with the results reported by Murata et al. The low strain rate poriton of the data can be represented by Equation (2) which describes the plastic properties of a variety of crystalline metals and alloys. The solid curve in Figure 1 is calculated by using Equation (2) with a value of log σ^* = 5.9 (σ^* in psi) and log $\dot{\epsilon}^*$ = -5.6.

The low stress dependence for plastic flow shown by the present data is a direct consequence of the high o* involved according to the deformation model. The value of o* can be used to estimate the flow stress at high strain rates for crystalline solids. 5 The value of o* (7.9x10⁵psi or 547 MPa) given above is considerably higher than the high flow stresses reported in the literature for Fe-Ni base metallic glasses. It is possible that the flow stress determination in a tensile test on metallic glasses will be lower than o* according to Equation (2). Based on recent work on work-hardening, the ability of a material to work harden is significantly reduced at high hardnesses (o*). 5 The lack of work-hardening suggested by the data in Figure 1 is therefore consistent with the high o* value given above. Essentially the present data suggest that we are observing here the properties of a highly workhardened crystalline solid. The deformation model will predict therefore little capacity for work hardening and a low stress dependence for plastic flow.

Figure 2 shows load relaxation data at 270°C obtained by using a specimen in the as-received condition without annealing at 300°C

prior to the load relaxation experiment. These $\log \sigma - \log \varepsilon$ curves have shapes similar to those given in Figure 1. However, the low strain rate portion (plastic part) of these curves do not coincide as well as those shown in Figure 1 reflecting the effects of structural rearrangement. Apparently annealing at 300°C improved the stability of the structure of the specimen. The improved stability found just below the annealing temperature has also been shown by Krenitsky and Ast 9 in their work on shear band deformation.

Figure 3 shows the load relaxation data obtained at room temperature by using as-received specimens. More than one specimen was used to obtain these curves. These curves show extremely high stress exponent (the inverse of the slope of the log σ - log $\dot{\epsilon}$ curve). By comparing with the curves in Figure 1, the room temperature curve suggests qualitatively the importance of anelastic deformation. Work is in progress in this laboratory to determine the flow laws for anelastic deformation in metallic glasses and to compare with those found in crystalline solids. 7 It should be noted that the data fit by using Equation (2) and shown in Figure 1 depends on the identification of the high strain rate portion of the log o - log & curve to be controlled by anelastic deformation. Information obtained on anelastic deformation together with the constitutive relations given previously can be used to account for the high strain rate portion of the log σ - log $\dot{\epsilon}$ data quantitatively in order to support the present analysis of the low strain rate portion of load relaxation data.

In summary the load relaxation data obtained in this work suggest the possibility that non-elastic deformation in metallic glasses is controlled by dislocation mechanisms. Both the anelastic component and the plastic component of deformation were identified

and the latter was found to be described by the same flow laws as those for plastic deformation in crystalline solids. The present data also supports the lack of substantial work hardening in metallic glasses.

This work is supported by Energy Research and Development Administration and by the Office of Naval Research.

REFERENCES

- 1. Lance A. Davis, Paper presented at the 1976 ASM Seminar on Metallic Glasses, Niagara Falls, N.Y., September 18-19. Proceedings to be published by American Society of Metals, 1977.
- T. Murata, H. Kimura and T. Masanoto, Scripta Met., Vol. 10, p. 705, (1976).
- 3. F. H. Huang, F. V. Ellis, C.Y. Li, Metallurgical Transactions A, Vol. 8, p. 699 (1977).
- 4. N. Nir, F. H. Huang, E. W. Hart, C. Y. Li, Metallurgical Transactions A, Vol. 8A, p. 583 (1977).
- 5. E. W. Hart, Che-Yu Li, H. Yamada, and G. L. Wire: "Phenomenological Theory: A Guide to Constitutive Relations and Fundamental Deformation Properties," in Constitutive Equations in Plasticity, A. Argon, ed., p. 149, MIT Press, (1975).
- 6. E. W. Hart, Acta. Met., Vol. 18, p. 599 (1970).
- 7. N. Nir, E. Hart, C. Y. Li, Scripta Metallurgica, Vol. 10, p. 189 (1976).
- 8. E. W. Hart, J. Eng. Mater. Technol., Vol. 98, p. 193 (1976).
- 9. D. Krenitsky and D. Ast, to be published.

FIGURE CAPTIONS

- Figure 1: Log σ Log $\dot{\epsilon}$ data for METGLAS 2826 obtained at 270°C after an 8 hr. anneal at 300°C.
- Figure 2: Log σ Log $\dot{\epsilon}$ data for METGLAS 2826 obtained at 270°C without preannealing.
- Figure 3: Log σ Log $\dot{\epsilon}$ data for METGLAS 2826 obtained at 20°C.

- T. Murata, H. Kimura and T. Masanoto, Scripta Met., 10, 705 (1976).
- 3. F. H. Huang, F. V. Ellis, C.Y. Li, Met. Trans. A, 8, 699 (1977).
- 1. N. Nir, F. H. Huang, E. W. Hart, C. Y. Li, Met. Trans. A, 8A, 583 (1977).
 - E. W. Hart, Che-Yu Li, W. Yamada, and G. L. Wire: "Phenomenological Theory: A Guide to Constitutive Relations and Fundamental Deformation Properties", in Constitutive Equations in Plasticity, A. Argon, ed., p. 149, MIT Press (1975).
- . E. W. Hart, Acta Met., 18, 599 (1970).
- 7. N. Nir, E. Hart, C. Y. Li, Scripta Met. 10, 189 (1976).
- 8. E. W. Hart, J. Engr. Mater. Technol., 98, 193 (1976).
- 9. D. Krenitsky and D. Ast, to be published.

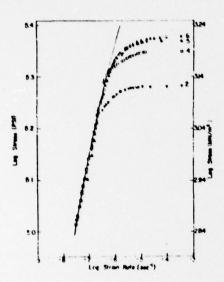


Figure 1

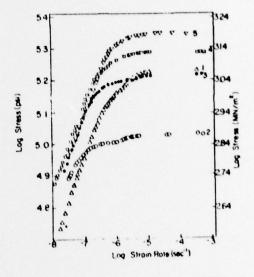


Figure 2

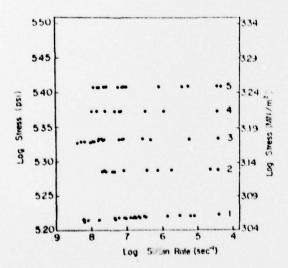


Figure 3

BASIC DISTRIBUTION 11ST

Technical and Summary Reports

April 1978

Organization	Copies	Organization	Copies
Defense Documentation Center Cameron Station Alexandria, VA 22314	12	Naval Air Propulsion Test Center Trenton, NJ 08628 ATTN: Library	r 1
Office of Naval Research Department of the Navy 800 N. Quincy Street Arlington, VA 22217		Naval Construction Batallion Civil Engineering Laboratory Port Hueneme, CA 93043 ATTN: Materials Division	1
ATTN: Code 471 Code 102 Code 470	1	Naval Electronics Laboratory San Diego, CA 92152 ATTN: Electron Materials Sciences Division	1
Commanding Officer Office of Naval Research Branch Office Building 114, Section D 656 Summer Street Boston, MA 02210	1	Naval Missile Center Materials Consultant Code 3312-1 Point Mugu, CA 92041	1
Commanding Officer Office of Naval Research Branch Office 536 South Clark Street Chicago, IL 60605	1	Commanding Officer Naval Surface Weapons Center White Oak Laboratory Silver Spring, MD 20910 ATTN: Library	1
Office of Naval Research OneHallidie Plaza Suite 601 San Francisco, CA 94102		David W. Taylor Naval Ship Research and Development Center Materials Department Annapolis, MD 21402	1
Naval Research Laboratory Washington, DC 20375	1	Naval Undersea Center San Diego, CA 92132 ATTN: Library	1
ATTN: Codes 6000 6100 6300 6400	1 1	Naval Underwater System Center Newport, RI 02840 ATTN: Library	1
2627 Naval Air Development Center Code 302	i	Naval Weapons Center China Lake, CA 93555 ATTN: Library	1
Warminstor, PA 18964 ATTN: Mr. F. S. Williams	1	Naval Postgraduate School Monterey, CA 53340 ATTN: Mechanical Engineering Department	1

	Organization	Copies	Organization	Copies
1	Naval Air Systems Command Washington, DC 20360 ATTN: Codes 52031		NASA Headquarters Washington, DC 20546 ATTN: Code:RRM	١
	52032	1	NASA	
,	Naval Sea System Command Washington, DC 20362 ATTN: Code 035	1	Lewis Research Center 21000 Brookpark Road Cleveland, OH 44135 ATTN: Library	١
1	Naval Facilities Engineering Command Alexandria, VA 22331 ATTN: Code 03	1	National Bureau of Standards Washington, DC 20234 ATTN: Metallurgy Division Inorganic Materials Div.	1
	Scientific Advisor Commandant of the Marine Corps Washington, DC 20380 ATTN: Code AX Waval Ship Engineering Center	1	Director Applied Physics Laboratory University of Washington 1013 Northeast Forthieth Street Seattle, WA 98105	y 1
1	Department of the Navy Washington, DC 20360 ATTN: Code 6101 Army Research Office	1	Defense Metals and Ceramics Information Center Battelle Memorial Institute 505 King Avenue Columbus, OH 43201	1
,	P.O. Box 12211 Iriangle Park, NC 27709 ATTN: Metallurgy & Ceramics Program Army Materials and Mechanics Research Center	1	Metals and Ceramics Division Oak Ridge National Laboratory P.O. Box X Oak Ridge, TN 37380 Los Alamos Scientific Laboratory	1
	Matertown, MA 02172 ATTN: Research Programs Office Air Force Office of Scientific	1	P.O. Box 1663 Los Alamos, NM 87544 ATTN: Report Librarian	1
	Research Bldg. 410 Bolling Air Force Base Washington, DC 20332 ATTN: Chemical Science Directorate Electronics & Solid State Sciences Directorate	1	Argonne National Laboratory Metallurgy Division P.O. Box 229 Lemont, IL 60439 Brookhaven National Laboratory Technical Information Division	١
W	Air Force Materials Laboratory Mright-Patterson AFB Dayton, OH 45433	1	Upton, Long Island New York 11973 ATTN: Research Library	1
• B	ibrary Building 50, Rm 134 Bawrence Radiation Laboratory Berkeley, CA	١	Office of Haval Research Branch Office 1030 East Green Street Pasadena, CA 91106	1

Professor G. S. Ansell Rensselaer Polytechnic Institute Department of Metallurgical Engineering Troy, NY 12181

Professor Dieter G. Ast Cornell University Department of Materials Science and Engineering Ithaca, NY 14853

Dr. E. M. Breinan United Technologies Corporation United Technologies Research Center East Hartford, CT 06108

Professor H. D. Brody University of Pittsburgh School of Engineering Pittsburgh, PA 14213

Dr. R. W. Cahn University of Sussex School of Engineering and Applied Science Brighton BN1 9QT ENGLAND

Ur. E. A. Clark
Solid State Division
Naval Surface Weapons Center
White Oak Laboratory
Silver Spring, MD 20910

Dr. S. M. Copley University of Southern California Los Angeles, CA 90007

Professor M. Cohen Massachusetts Institute of Technology Department of Metallurgy Cambridge, MA 02139 Dr. R. B. Diegle Battelle 505 King Avenue Columbus, OH 43201

Professor B. C. Giessen Northeastern University Department of Chemistry Boston, MA 02115

Professor N. J. Grant
Massachusetts Institute of Technology
Department of Materials Science
and Engineering
Cambridge, MA 02100

Dr. F. E. Luborsky General Electric Company P. O. Box 8 Corporate R&D Schenectady, NY 12301

Dr. J. Perel Phrasor Technology 1536 Highland Avenue Duarte, CA 91010

Professor O. D. Sherby Stanford University Materials Science Division Stanford, CA 94300

Professor D. Turnbull Harvard University Division of Engineering and Applied Physics Cambridge, MA 02138

Professor R. Mehrabian University of Illinois Department of Mechanical and Industrial Engineering Urbana, IL 61801

Professor P. R. Strutt University of Connecticut School of Engineering Department of Latallurgy Storrs, CT 06268